

Rate of change of Lithium moles

$$\frac{d}{dt} \int_{\mathcal{R}} c \rho \, dV = \frac{d}{dt} \int_{\mathcal{R}_0} \underbrace{c \rho \det F_0}_{\rho_0} \, dV \quad F_0 \equiv \nabla \phi$$

$$= \int_{\mathcal{R}_0} (\dot{c} \rho + c \dot{\rho} + c \rho \operatorname{div} v) \det F_0 \, dV$$

$$= \int_{\mathcal{R}_0} \dot{c} \rho \det F_0 \, dV + \int_{\mathcal{R}_0} \underbrace{c (\dot{\rho} + \rho \operatorname{div} v)}_0 \det F_0 \, dV$$

$$= \int_{\mathcal{R}_0} \dot{c} \rho_0 \, dV = \int_{\mathcal{R}} \dot{c} \rho \, dV$$

[Notebook page scanned on 2016/06/30]

Lithium flux

$$\int_{\partial \mathcal{R}} h \cdot n \, dA = \int_{\partial \mathcal{R}_0} (h \cdot n) \frac{A_F}{A_{F_0}} \, dA = \int_{\partial \mathcal{R}_0} h \cdot (\text{col } F_0) m_0 \, dA$$

$$= \int_{\partial \mathcal{R}_0} h \cdot F_0^{-T} m_0 (\det F_0) \, dA$$

$$= \int_{\partial \mathcal{R}_0} \underbrace{(\det F_0) F_0^{-1}}_{h_0} h \cdot m_0 \, dA$$

$$= \int_{\partial \mathcal{R}_0} h_0 \cdot m_0 \, dA$$

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Species molar balance

$$\frac{d}{dt} \int_{\mathcal{R}} c_p dV = - \int_{\partial \mathcal{R}} h \cdot n d\Delta + \int_{\mathcal{R}} h dV$$

flux source

We assume $h=0$ (no production/reaction)

$$\int_{\mathcal{R}} \dot{c}_p dV = - \int_{\partial \mathcal{R}} h \cdot n d\Delta$$

$$\int_{\mathcal{R}_0} \dot{c}_{p_0} dV = - \int_{\partial \mathcal{R}_0} h_0 \cdot n_0 d\Delta$$

which localize to

$$\dot{c}_p + \operatorname{div} h = 0 \quad \text{in } \mathcal{R}$$

$$\dot{c}_{p_0} + \operatorname{div} h_0 = 0 \quad \text{in } \mathcal{R}_0$$

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From molar balance to power balance

let us introduce a scalar field (chemical potential)

$$\mu_0(x) = \mu(\phi(x))$$

"energy per mole of lithium"

in order to turn the species molar balance into a power molar balance:

$$\int_{\mathcal{R}} (\dot{c}_p + \operatorname{div} h) \mu \, dV = 0 \quad \forall \mu$$

$$\int_{\mathcal{R}_0} (\dot{c}_{p_0} + \operatorname{div} h_0) \mu_0 \, dV = 0 \quad \forall \mu_0$$

$$\Rightarrow \int_{\mathcal{R}_0} \mu_0 \dot{c}_{p_0} \, dV = - \int_{\mathcal{R}_0} \operatorname{div} (\mu_0 h_0) \, dV + \int_{\mathcal{R}_0} h_0 \cdot \nabla \mu_0 \, dV = 0$$

$$\int_{\mathcal{R}_0} \mu_0 \dot{c}_{p_0} \, dV = - \int_{\partial \mathcal{R}_0} \mu_0 h_0 \cdot n_0 \, dA + \int_{\mathcal{R}_0} h_0 \cdot \nabla \mu_0 \, dV = 0$$

with $\operatorname{div} (\mu_0 h_0) = \operatorname{tr} (\nabla (\mu_0 h_0)) = \operatorname{tr} (\mu_0 \nabla h_0 + \nabla \mu_0 \otimes h_0)$
 $= \mu_0 \operatorname{div} h_0 + h_0 \cdot \nabla \mu_0$

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Power balance laws

$$\int_{\mathcal{R}_0} b_0 \cdot \tau_0 dV + \int_{\partial \mathcal{R}_0} t_0 \cdot \tau_0 dA = \int_{\mathcal{R}_0} S_0 \cdot \nabla \tau_0 dV \quad \nabla \tau_0$$

external power expenditure [→ G. (27.20)]

$$\int_{\mathcal{R}_0} \mu_0 \dot{c} \rho_0 dV = - \int_{\partial \mathcal{R}_0} \mu_0 h_0 \cdot n_0 dA + \int_{\mathcal{R}_0} h_0 \cdot \nabla \mu_0 dV \quad \nabla \mu_0$$

external power expenditure

Free energy imbalance principle:

for any constitutive process (for any realizable evolution)

$$S_0 \cdot \dot{F}_0 + \mu_0 \dot{c} \rho_0 - h_0 \cdot \nabla \mu_0 - \frac{d}{dt} \psi \geq 0$$

[very general expression]

Let us assume

$$\psi = \hat{\psi}(F_0, c)$$

and

$$\frac{d}{dt} \hat{\psi}(F_0, c) = S_0 \cdot \dot{F}_0 + \mu_0 \rho_0 \dot{c}$$

Fick's law

Because of the previous assumption the free energy imbalance reduces to

$$-h_o \cdot \nabla \mu_o \geq 0$$

which must be fulfilled in any realizable evolution.

This requirement leads to the following characterization of the molar flux

$$h_o = -M \nabla \mu_o \quad \text{Fick's law}$$

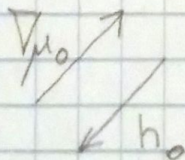
with M a positive definite tensor (mobility)

$$M \nabla \mu_o \cdot \nabla \mu_o \geq 0$$

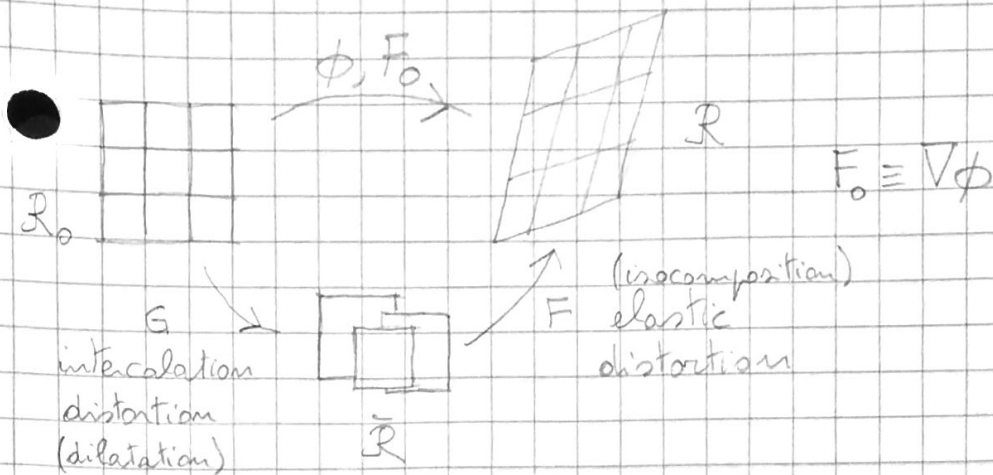
As a special case we get

$$h_o = -d \nabla \mu_o$$

with $d \geq 0$ (diffusion coefficient)



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$$\beta = \det G \quad G = \beta^{1/3} I$$

$$F_0 = FG$$

stoichiometric constant α

$$\det G = (\rho_0 V_0 + \rho_{Li} V_{Li}) / (\rho_0 V_0) = 1 + c \frac{V_{Li}}{V_0} = 1 + c \alpha$$

$$\dot{\beta} = \alpha \dot{c} \quad \frac{d}{dt}(\det G) = (\det G) \operatorname{tr}(\dot{G} G^{-1}) = \dots = \alpha \dot{c}$$

$$\dot{G} G^{-1} = \frac{1}{3} \beta^{-2/3} \beta^{-1/3} \alpha \dot{c} I = \frac{1}{3} \frac{\alpha}{\beta} \dot{c} I$$

$$\dot{F}_0 F_0^{-1} = (\dot{F} G + F \dot{G}) G^{-1} F^{-1} = \dot{F} F^{-1} + \frac{1}{3} \frac{\alpha}{\beta} \dot{c} I$$

$$S_0 = T F_0^{-T} (\det F_0) \quad S = T F^{-T} (\det F)$$

$$S_0 \cdot \dot{F}_0 = (\det F_0) T F_0^{-T} \cdot \dot{F}_0 = \beta (\det F) T \cdot \dot{F}_0 F_0^{-1}$$

$$= \beta (\det F) T \cdot \dot{F} F^{-1} + \frac{1}{3} \alpha (\det F) T \cdot I \dot{c}$$

$$= \beta S \cdot \dot{F} + \frac{1}{3} \alpha (\det F) \operatorname{tr} T \dot{c}$$

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Hence, according to the previous assumptions about the free energy, we get

$$\begin{aligned} \frac{d}{dt} \hat{\psi}(F_0, c) &= S_0 \cdot \dot{F}_0 + \mu_0 \dot{\rho}_0 c \\ &= \beta S \cdot \dot{F} + \left(\frac{1}{3} \alpha (\det F) \operatorname{tr} T + \mu_0 \rho_0 \right) \dot{c} \end{aligned}$$

If we further assume for the free energy the expression

$$\hat{\psi}(F_0, c) = (\det G) \varphi(F)$$

with

$$\frac{d}{dt} \varphi(F) = S \cdot \dot{F}$$

we finally get

$$\begin{aligned} \beta \frac{d}{dt} \varphi(F) + \dot{\beta} \varphi(F) \\ = \beta S \cdot \dot{F} + \left(\frac{1}{3} \alpha (\det F) \operatorname{tr} T + \mu_0 \rho_0 \right) \dot{c} \end{aligned}$$

$$\left(\frac{1}{3} \alpha (\det F) \operatorname{tr} T - \alpha \varphi(F) + \mu_0 \rho_0 \right) \dot{c} = 0$$

$$\Rightarrow \mu_0 = \frac{\alpha}{\rho_0} \left(\varphi(F) - \frac{1}{3} (\det F) (\operatorname{tr} T) \right)$$

We could start directly from the free energy

expression

$$\psi = (\det G) \varphi(F) \quad \det G = \beta$$

$$\frac{d}{dt} \psi = \dot{\beta} \varphi(F) + \beta \frac{d}{dt} \varphi(F) = \dot{\beta} \varphi(F) + \beta \hat{S}(F) \cdot \dot{F}$$

$$\dot{F}_0 = \frac{d}{dt} (F \beta^{1/3}) = \beta^{1/3} \dot{F} + \frac{1}{3} \beta^{-2/3} \dot{\beta} F$$

$$\beta^{2/3} \dot{F}_0 = \beta \dot{F} + \frac{1}{3} \dot{\beta} F$$

$$\beta \hat{S}(F) \cdot \dot{F} = \hat{S}_0(F) \cdot \left(\beta^{2/3} \dot{F}_0 - \frac{1}{3} \dot{\beta} F \right)$$

$$S_0 = (\det G) S G^{-T} = \beta S \beta^{-1/3} = \beta^{2/3} S$$

$$\beta \hat{S}(F) \cdot \dot{F} = \hat{S}_0(F) \cdot \dot{F}_0 - \frac{1}{3} \hat{S}(F) F^T \cdot I \dot{\beta}$$

$$S = (\det F) T F^{-T} \Rightarrow \hat{S}(F) F^T \cdot I = (\det F) \operatorname{tr} \hat{T}(F)$$

$$\beta \hat{S}(F) \cdot \dot{F} = \hat{S}_0(F) \cdot \dot{F}_0 - \frac{1}{3} (\det F) \operatorname{tr} \hat{T}(F) \dot{\beta}$$

We get also

$$\beta S \cdot \dot{F} = S_0 \cdot \dot{F}_0 - \frac{1}{3} (\det F) (\operatorname{tr} T) \dot{\beta}$$

[2015-11-16]

From the free energy imbalance

$$S_0 \cdot \dot{F}_0 + \mu_0 \rho_0 \dot{c} - h_0 \cdot \nabla \mu_0 - \frac{d}{dt} \psi \geq 0$$

we get

$$\beta S \cdot \dot{F} + \frac{1}{3} (\det F) (\text{tr} T) \dot{\beta} + \mu_0 \rho_0 \dot{c} - h_0 \cdot \nabla \mu_0 - \varphi \dot{\beta} - \beta \hat{S}(F) \cdot \dot{F} \geq 0$$

Rearranging terms

$$\beta (S - \hat{S}(F)) \cdot \dot{F} + \left(\frac{1}{3} (\det F) (\text{tr} T) - \varphi \right) \dot{\beta} + \mu_0 \rho_0 \dot{c} - h_0 \cdot \nabla \mu_0 \geq 0$$

Let us set now $\beta = 1 + \alpha c$

This relationship between concentration and volume change is a stoichiometric relation about the intercalation of atoms in a lattice cell, when describing the diffusion of a species in a lattice.

In poroelasticity the relation above is not being but the saturation condition.

Further, relying on the assumption

$$\beta = 1 + \alpha c$$

we can introduce a new balance law

$$\int_{B_0} (\check{\rho}_0 \dot{\beta} - \mu_0 \rho_0 \dot{c}) dV = 0 \quad \forall \dot{c}$$

relating the power expended by an interstitial measure to the rate of change of the energy transported by a diffusing species.

The local form of the balance law is

$$\alpha \check{\rho}_0 = \mu_0 \rho_0 \quad \Rightarrow \quad \mu_0 = \frac{\alpha}{\rho_0} \check{\rho}_0$$

Denoting by J and p $J = \det F$, $p = -\frac{1}{3}(\text{tr} T)$,

the free energy imbalance takes the form

$$\underbrace{\beta (S - \hat{S}(F)) \cdot \dot{F}}_{S^+} + \underbrace{(-Jp - \psi + \check{\rho}_0)}_{\check{\rho}_0^+} \dot{\beta} - \frac{\alpha}{\rho_0} h_0 \cdot \nabla \check{\rho}_0 \geq 0$$

or, equivalently,

$$\beta S^+ \cdot \dot{F} + \check{\rho}_0^+ \dot{\beta} - \frac{\alpha}{\rho_0} h_0 \cdot \nabla \check{\rho}_0 \geq 0$$

[2016-11-18]

with

$$S = \hat{S}(F) + S^+$$

$$\check{p}_0 = \varphi + Jp + \check{p}_0^+$$

characterizing the stress and the interstitial pressure through S^+ and \check{p}_0^+ , dissipative terms subject, in any process to the dissipation inequality above,

As a consequence we obtain as separate conditions

$$S^+ \cdot \dot{F} \geq 0$$

$$\check{p}_0^+ \dot{\beta} \geq 0 \quad (\dot{\beta} \text{ is independent of } \dot{F})$$

$$-\frac{\alpha}{p_0} h_0 \cdot \nabla \check{p}_0 \geq 0 \quad (\text{to cover the case where } \dot{\beta}=0, \dot{F}=0)$$

Since

$$S^+ \cdot \dot{F} = S^+ F^T \cdot \dot{F} F^{-1} = J^{-1} T^+ \cdot \dot{F} F^{-1}$$

a standard choice is

$$T^+ = 2\mu_s \text{sym}(\dot{F} F^{-1}) \quad \text{with } \mu_s \geq 0$$

leading to

$$T^+ \cdot \dot{F} F^{-1} \geq 0$$

Similarly we can choose

$$\check{p}_0^+ = \mu_v \dot{\beta} \quad \text{with } \mu_v \geq 0$$

- In order to fulfil the last inequality we arrive at the Darcy's law

$$h_0 = -M \nabla \mu_0 = -M \frac{\alpha}{P_0} \nabla P_0$$

as a variant of the Fick's law

- resulting from the special choice of the free energy expression.

Denoting by V^m the molar volume of the interstitial species in the reference shape we may define the discharge per unit reference area

$$q_0 = h_0 V^m$$

- and get

$$q_0 = -V^m \frac{\alpha}{P_0} M \nabla P_0$$

The corresponding units of measure turn out to be

$$\begin{aligned} [q_0] &= (\text{m}^3 \text{mol}^{-1}) (\text{mol}^{-1} \text{m}^3) (\text{s}^{-1} \text{mol}^2 \text{Pa}^{-1} \text{m}^{-4}) (\text{Pa} \text{m}^{-1}) \\ &= \text{m}^6 \text{mol}^{-2} \text{s}^{-1} \text{mol}^2 \text{m}^{-5} = \text{s}^{-1} \text{m} \end{aligned}$$

● The total discharge is $Q_0 = q_0 A \Rightarrow [Q_0] = \text{m}^3 \text{s}^{-1}$

Unit of measure

$$[h] = \text{mol m}^{-2} \text{s}^{-1}$$

$$[\mu] = \text{J mol}^{-1}$$

$$[\mu c_p] = \text{J mol}^{-1} \text{s}^{-1} \text{mol m}^{-3} = \text{J mol}^{-3} \text{s}^{-1}$$

$$[h \nabla \mu] = (\text{mol m}^{-2} \text{s}^{-1}) (\text{J mol}^{-1} \text{m}^{-1}) = \text{J m}^{-3} \text{s}^{-1}$$

$$h = -M \nabla \mu$$

$$\Rightarrow [M] = (\text{mol m}^{-2} \text{s}^{-1}) (\text{J mol}^{-1} \text{m}^{-1})^{-1} = \text{J}^{-1} \text{mol}^2 \text{m}^{-1} \text{s}^{-1}$$

molar volume

$$[V^m] = \text{m}^3 \text{mol}^{-1}$$

discharge

$$[q] = (\text{m}^3 \text{mol}^{-1}) (\text{mol m}^{-2} \text{s}^{-1}) = \text{m s}^{-1}$$

amount of matter measured by mol

As an alternative it can be measured by kg

[2016-12-17]